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9-(1,3-Anhydro- β -D-psicofuranosyl)adenine

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In the structure of the title compound, $C_{11}H_{13}N_5O_4$, the glycosidic torsion angle, χ , is $-107.1 (2)^{\circ}$ [nucleic acid nomenclature used throughout the manuscript; IUPAC-IUB Joint Commision on Biochemical Nomenclature (1983). *Eur. J. Biochem.* **131**, 9–15], indicating the *anti* conformation. The furanosyl ring adopts an N-type sugar pucker with the following pseudorotational parameters: $P_N = 50.5 (2)^{\circ}$ and $\nu_{max} = 34.9 (1)^{\circ}$. The conformation around the C5' – C6' bond is *ap* (*gauche,trans*; *gt*; –*g*), with a torsion angle γ of 176.28 (19)°. The 1',2'-oxetane ring is not planar but folded along C3'...C1', with an angle of 9.6 (1)°.

Comment

In recent years, conformationally rigid 1,3-anhydro- β -Dpsicofuranosyl nucleosides have attracted much attention as constituents of oligonucleotides, the exciting biochemical and biophysical properties of which have been investigated in detail by Chattopadhyaya and co-workers (Bogucka *et al.*, 2005, and references therein); however, structural data have not heretofore been published. Recently, we described the synthesis of the anhydro nucleoside (I) (Roivainen *et al.*, 2002), and deduced its structure from UV and NMR spectroscopic data to be consistent with that shown in the scheme below. Its single-crystal X-ray structure (Fig. 1) was determined in order to confirm this assignment.



As might be expected, the structures of the adenine bases of (I) and adenosine (II) (Lai & Marsh, 1972), which represents the conformationally unrigid counterpart to (I), are very similar. The orientation of the almost planar adenine base of (I) relative to the sugar ring is *anti*, with a glycosyl torsion

angle χ (C4-N9-C2'-O5') (IUPAC-IUB Joint Commission on Biochemical Nomenclature, 1983) of $-107.1 (2)^{\circ}$, which differs substantially from that of -170.1° found for adenosine. The furanosyl ring of (I) in the solid state adopts the N-type sugar pucker with the following pseudorotational parameters: $P_{\rm N} = 50.5 \ (2)^{\circ} \ (\text{C5'-exo})$ and $v_{\rm max} = 34.9 \ (1)^{\circ}$. The N conformation was also found in the crystal structure of adenosine, although through somewhat different pseudorotational parameters, viz. $P_{\rm N} = 7.2^{\circ} (C2' - exo/C3' - endo; {}^{3}T_{2})$ and $v_{\rm max} = 36.0^{\circ}$ (Lai & Marsh, 1972). It is noteworthy that the insertion of the 1',2'-oxetane ring into adenosine did not lead to an essential change of bond distances within the furanose ring (the deviations are ± 0.007 Å for all bonds, but the C5'-C4' bond was shorter by 0.014 Å than the relevant C3'-C4' bond of adenosine). Thus, the C5'-O5' bond is longer than O5'-C2', in accordance with an analogous correlation in most purine nucleosides (Seela et al., 1999). The 1',2'-oxetane ring itself is not planar but folded along $C3' \cdots C1'$, with an angle of 9.6 (1)°.

The glycosidic bond length of (I) [C2'-N9 = 1.438 (2) Å] is shorter than that of adenosine by 0.028 Å (Lai & Marsh, 1972). The conformations around the exocyclic C5'-C6' bond in the solid state of (I) and the corresponding C4'-C5' bond of adenosine are similar, *viz. ap* (gauche,trans; gt; -g), with torsion angles of 176.28 (19) and 177.0°, respectively.

In the extended structure, all molecules are linked together via a three-dimensional network of hydrogen bonds. The main intermolecular feature, an eight-membered ring (Fig. 2), consists of hydroxy groups O6' and O4' of two different molecules, and amino group N6 and atom N1 of a third molecule. Whereas the hydroxy groups function as both donors and acceptors of hydrogen bonds, atom N1 serves only as an acceptor and the NH₂- group as a donor. Three of the four intermolecular hydrogen bonds are found within this ring system. The fourth connects a fourth, exocyclic, molecule with the NH₂ groups inside the eight-membered supramolecular ring. The geometric details of all four hydrogen bonds are given in Table 2. As can be seen from the donor-acceptor distances, which are in the range 2.633 (3)-3.043 (2) Å, only the hydrogen bond between the hydroxy groups is relatively strong, whereas the other three are weaker.



Figure 1

A perspective view of nucleoside (I), showing the atom-numbering scheme, displacement ellipsoids at the 50% probability level for non-H atoms and H atoms as spheres of small arbitrary size.



Figure 2

Part of the crystal structure of nucleoside (I), showing the main structural features of the hydrogen-bonding scheme. [Symmetry codes for the generation of the different molecules are as follows: (1) x, y, z; (2) 2 - x, $-\frac{1}{2} + y$, 1 - z; (3) 1 + x, y, 1 + z; (4) 1 - x, $\frac{1}{2} + y$, 1 - z.]

Experimental

The synthesis of compound (I) has been described previously (Roivainen *et al.*, 2002). Samples for X-ray analyses were crystallized from a mixture of methanol and 2-propanol. Single crystals suitable for X-ray diffraction were selected directly from the sample as prepared.

Crystal data

 $C_{11}H_{13}N_5O_4$ $M_r = 279.26$ Monoclinic, $P2_1$ a = 5.4154 (5) Å b = 9.8941 (8) Å c = 11.4431 (12) Å $\beta = 94.970 (14)^{\circ}$ $V = 610.82 (10) Å^3$

Data collection

Bruker P4 diffractometer ω scans 2053 measured reflections 1878 independent reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.041$ $wR(F^2) = 0.109$ S = 1.041878 reflections 186 parameters H atoms treated by a mixture of independent and constrained refinement $w = 1/[\sigma^2(F_o^2) + (0.0627P)^2 + 0.0578P]$ where $P = (F_o^2 + 2F_o^2)/3$ Z = 2 $D_x = 1.518 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation $\mu = 0.12 \text{ mm}^{-1}$ T = 293 (2) K Plate, colourless $0.35 \times 0.22 \times 0.08 \text{ mm}$

1686 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.023$ $\theta_{\text{max}} = 30.0^{\circ}$

 $\begin{array}{l} (\Delta/\sigma)_{\rm max} < 0.001 \\ \Delta\rho_{\rm max} = 0.24 \ {\rm e} \ {\rm \AA}^{-3} \\ \Delta\rho_{\rm min} = -0.23 \ {\rm e} \ {\rm \AA}^{-3} \\ {\rm Extinction \ correction: } SHELXTL \\ {\rm Extinction \ coefficient: } 0.037 \ (8) \\ {\rm Absolute \ structure: \ based \ on \ known \\ absolute \ configuration \ of \ the \\ chemical \ entity \\ {\rm Flack \ parameter: } -0.7 \ (13) \end{array}$

In the absence of suitable anomalous scattering atoms, refinement of the Flack (1983) parameter led to an inconclusive result. All H atoms were initially found in a difference Fourier synthesis. In order to maximize the data/parameter ratio, H atoms bonded to C atoms were placed in geometrically idealized positions (C-H = 0.93– 0.98 Å) and constrained to ride on their parent atoms. In order to describe the hydrogen-bonding scheme as well as possible, the positions of the H atoms of the OH and NH₂ groups were first allowed to refine restrained to common O-H and N-H bond lengths (DFIX).

Table 1

Selected geometric parameters (Å, °).

N9-C2′	1.438 (2)	C2′-O5′	1.418 (2)
C1'-O3'	1.463 (3)	C2'-C3'	1.533 (3)
C1'-C2'	1.535 (3)	C3'-O3'	1.451 (3)
O4′-C4′	1.409 (2)	C5′-O5′	1.450 (2)
C8-N9-C2'	122.95 (17)	O3'-C3'-C4'	111.13 (16)
O5'-C2'-N9	106.95 (17)	O3'-C3'-C2'	90.96 (15)
N9-C2'-C3'	119.45 (16)	C3'-O3'-C1'	91.80 (15)
O5'-C2'-C1'	116.03 (16)	O5'-C5'-C4'	104.89 (16)
N9-C2'-C1'	119.60 (17)	C3'-C4'-C5'	102.24 (15)
C3'-C2'-C1'	86.02 (16)	C2'-O5'-C5'	108.25 (15)
C2-N3-C4-N9	-176.7 (2)	C3'-C2'-O5'-C5'	-18.9 (2)
N7-C5-C6-N6	-2.2(4)	C1'-C2'-O5'-C5'	75.3 (2)
N3-C4-N9-C8	-179.9(2)	C6'-C5'-O5'-C2'	155.3 (2)
C4-N9-C2'-O5'	-107.1(2)	C4′-C5′-O5′-C2′	33.1 (2)
N9-C2'-O5'-C5'	-148.35 (16)	C4'-C5'-C6'-O6'	176.28 (19)

Table 2			
Hydrogen-bond	geometry	(Å,	°).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N6-H61\cdots O3'^{i}$ $N6-H62\cdots O4'^{ii}$ $O4'-H4'O\cdots O6'^{iii}$ $O6'-H6'O\cdots N1^{iv}$	0.87 0.87 0.91 0.91	2.09 2.22 1.75 1.87	2.932 (3) 3.043 (2) 2.633 (3) 2.782 (3)	166 159 162 175

Symmetry codes: (i) -x + 1, $y + \frac{1}{2}$, -z + 1; (ii) x + 1, y, z + 1; (iii) -x + 1, $y - \frac{1}{2}$, -z; (iv) -x + 2, $y + \frac{1}{2}$, -z + 1.

After refinement, the positions of these H atoms were also constrained (AFIX 3) to ride on their parent atoms.

Data collection: *XSCANS* (Siemens, 1996); cell refinement: *XSCANS* (Siemens, 1996); data reduction, structure solution and definement, molecular graphics and preparation of publication material: *SHELXTL* (Sheldrick, 1997).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: FA3039). Services for accessing these data are described at the back of the journal.

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